

## **AMENDMENTS TO THE CLAIMS:**

The listing of claims will replace all prior versions, and listings of claims in the application:

### **LISTING OF THE CLAIMS**

1. (Currently Amended) A battery ~~(1, 138, 158, 180, 200)~~ comprising:  
an anode ~~(10, 140, 160, 182, 202)~~; <sub>i</sub>  
a cathode ~~(12, 146, 168, 184, 208)~~; <sub>i</sub> and  
an electrolyte ~~(20, 148)~~ therebetween, ~~the battery characterized by:~~ at least one  
of the anode ~~(10, 140, 160, 182, 202)~~ and cathode ~~(12, 184, 208)~~ comprising an  
electrically conductive sponge material ~~(16, 114, 164, 204)~~ , the electrically conductive  
sponge material defining dendrites having a width of less than 1 micrometer.
2. (Currently Amended) The battery of claim 1, ~~further characterized by:~~  
wherein the at least one of the anode and cathode ~~comprising comprises~~ an electrically  
conductive substrate ~~(14, 82, 102, 120, 142, 162, 186, 188, 206)~~ which is in electrical  
contact with the electrically conductive sponge material.
3. (Currently Amended) The battery of claim 2, ~~further characterized by~~  
wherein the substrate ~~being is~~ in the form of a foil ~~(14)~~, wire, ribbon ~~(98, 102, 120)~~, cast  
structure, or sheet ~~(14, 82)~~.
4. (Currently Amended) The battery of claim 2, ~~further characterized by~~  
wherein the substrate and the electrically conductive sponge ~~being are~~ formed from the  
same material.
5. (Currently Amended) The battery of claim 2, ~~further characterized by~~  
wherein the substrate ~~including includes~~ a metal selected from the group consisting of  
silver, copper, and aluminum.
6. (Currently Amended) The A battery ~~of claim 2, further characterized by~~  
comprising:  
an anode;

a cathode; and

an electrolyte therebetween, at least one of the anode and cathode comprising a substrate and an electrically conductive sponge material, the sponge material being in the form of particles of sponge attached to the substrate.

7. (Currently Amended) The A battery of claim 2, further characterized by comprising:

an anode

a cathode; and

an electrolyte therebetween, at least one of the anode and cathode comprising a substrate and an electrically conductive sponge material, the sponge material being in the form of a layer of sponge grown on the substrate.

8. (Currently Amended) The battery of claim 1, further characterized by wherein the at least one of the anode and cathode ~~comprising~~ comprises a plurality of thin layers of the electrically conductive sponge material.

9. (Currently Amended) The battery of claim 1, further characterized by wherein the electrically conductive sponge material ~~including~~ includes an element selected from the group consisting of copper, silver, gold, aluminum, and combinations thereof.

10. (Currently Amended) The battery of claim ~~[[1]]~~ 7, further characterized by wherein the electrically conductive sponge material ~~defining~~ defines dendrites.

11. (Currently Amended) The battery of claim 10, further characterized by wherein the dendrites ~~having~~ have a width of from about 200 nanometers to 30 micrometers.

12. (Currently Amended) The battery of claim 10, further characterized by wherein the at least one of the anode and cathode ~~including~~ includes at least one substrate layer and wherein the dendrites extend from the substrate.

13. (Currently Amended) The battery of claim 12, further characterized by

wherein the dendrites ~~(18)~~ extending extend generally perpendicularly from the substrate layer.

14. (Currently Amended) The battery of ~~any one of~~ claim 1, ~~further characterized by~~ wherein the sponge material ~~being~~ is coated with an electrically conductive material.

15. (Withdrawn, Currently Amended) A method of ~~conducting a reaction on a surface~~ forming the battery of claim 7, the method comprising:

~~forming the surface (36);~~

~~contacting the surface with a reactant[[]]; and~~

~~allowing the reaction to take place, the method characterized by: the step of forming the surface including:~~

~~growing a sponge material (16, 114, 164, 204) having a plurality of open pores (37, 130) which are accessible to the reactant~~ electrolyte; and

forming the at least one of the anode and the cathode from the grown sponge.

16. (Withdrawn, Currently Amended) The method of claim 15, ~~further characterized by~~ wherein the sponge material ~~including~~ includes one of the group consisting of Li, Be, Mg, Al, Si, Ca, Sc, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Ge, Sr, Y, Zr, Nb, Mo, Rh, Pd, Ag, Cd, In, Sn, Sb, Te, Ba, La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Hf, Ta, W, Rh, Os, Ir, Pt, Au, Tl, Pb, Bi, Po, and combinations thereof.

17. (Withdrawn, Currently Amended) The method of claim 15, ~~further characterized by~~ wherein the step of growing the sponge ~~including~~ includes:

~~reducing a compound to its elemental form, the elemental form occupying a smaller volume than the compound such that an open porous structure is formed.~~

18. (Withdrawn, Currently Amended) The method of claim 17, ~~further characterized by~~ wherein the step of growing the sponge ~~including~~ includes:

~~heating a substrate (14) to a deposition temperature;~~

~~contacting the substrate with a vapor which includes a halide of the sponge material in the presence of an alkali or earth alkali metal vapor, the sponge halide vapor~~

reacting with the alkali or earth alkali metal vapor to form the sponge material and an alkali or earth alkali metal halide, the sponge material being deposited on the substrate in the form of dendrites ~~(18)~~.

19. (Withdrawn, Currently Amended) The method of claim 18, further ~~characterized by~~ comprising:

removing the alkali or earth alkali metal halide from the sponge material by vacuum distillation.

20. (Withdrawn, Currently Amended) The method of claim 15, further ~~characterized by~~ wherein the step of growing the sponge ~~including~~ includes:

heating the sponge material ~~(86, 114)~~ with an insoluble material ~~(88, 116)~~ to a temperature at which the sponge material and the insoluble material are both liquids, the sponge material and the insoluble material being mutually insoluble at a temperature at which the sponge material freezes;

mixing the two liquids;

cooling the sponge material and the insoluble material to form the sponge; and removing the insoluble material from the sponge.

21. (Withdrawn, Currently Amended) The method of claim 20, further ~~characterized by~~ wherein the step of cooling ~~including~~ includes:

cooling the sponge material ~~(86)~~ and the insoluble material ~~(88)~~ adjacent a substrate ~~(82)~~ to form a directionally grown sponge of the sponge material on the substrate.

22. (Withdrawn, Currently Amended) The method of claim 20, further ~~characterized by~~ wherein the sponge material ~~comprising~~ comprises an element selected from the group consisting of Mg, Al, Si, Zn, Ga, Ge, As, Se, Cd, In, Sn, Sb, Cv, Ni, Ag, Ti, Te, Tl, Pb, Bi, and alloys thereof.

[[22]] 23. (Withdrawn, Currently Amended) The method of claim 21, further ~~characterized by~~ wherein the insoluble substance ~~being~~ is selected from the group consisting of Na, K, Rb, Cs, Ca, Sr, Ba, and salts thereof.

[[23]] 24. (Withdrawn, Currently Amended) The method of claim 17 ~~further characterized by wherein~~ the step of growing the sponge material ~~including includes~~:

forming an oxide scale ~~(122)~~ on a substrate ~~(120)~~, the substrate including a metal which is oxidizable to an oxide having a lower density than the substrate, by oxidizing at least an outer portion ~~(124)~~ of the substrate to form the oxide scale; and reducing the oxide scale to the metal, the metal having an open porous structure.

[[24]] 25. (Withdrawn, Currently Amended) The method of claim [[23]] 24, ~~further characterized by wherein~~ the step of reducing ~~including includes~~:

reducing the oxide with a reducing agent ~~(128)~~; and  
the method further including after the step of reducing:  
removing an oxide of the reducing agent from the porous metal.

[[25]] 26. (Withdrawn, Currently Amended) The method of claim [[24]] 25, wherein the oxide of the reducing agent is a fluid.

[[26]] 27. (Withdrawn, Currently Amended) A battery comprising:

an anode;

a cathode;

an electrolyte therebetween, at least one of the anode and cathode comprising an electrically conductive metal sponge, the metal sponge having with high specific surface area that is accessible to reactants with reaction paths that are no larger than two times the sponge thickness.

[[27]] 28. (Withdrawn) A metal sponge with a geometry of open porosity between dendrites that enables through-flow of gas or liquid with an electrical resistance that is at least two times less than that of an ordinary sintered powder sponge.

29 (New) A battery comprising:

an anode;

a cathode;

an electrolyte therebetween, at least one of the anode and cathode comprising an electrically conductive sponge material the electrically conductive sponge material

defining dendrites; and

a layer of a different material on the dendrites which spaces the dendrites from the electrolyte.

30. (New) The battery of claim 29 and wherein the sponge defines pores having a pore width of less than 30 micrometers, the layer of a different material covering the sides of the pores to provide a passage for access of the electrolyte.

31. (New) The battery of claim 29, wherein the dendrites have a width of less than 30 micrometers in thickness.

32. (New) A battery comprising:

an anode;

a cathode;

an electrolyte therebetween, at least one of the anode and cathode comprising an electrically conductive sponge material the electrically conductive sponge material defining dendrites, the sponge having a specific surface area of at least  $40\text{m}^2/\text{cm}^3$ .

**Amendments to the Drawings**

Two drawing sheets showing FIGURES 2 and 11A, 11B 11C, and 11D are attached.